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3-Arylidene-pent-2,4-dione in the Synthesis of Heterocyclic Systemes. Facile One-Pot Synthesis of Novel Pentasubstituted 1,8-Naphthyridine-2-thione

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3-Arylidene-pent-2,4-dione in the Synthesis of Heterocyclic Systemes. Facile One-Pot Synthesis of Novel Pentasubstituted 1,8-Naphthyridine-2-thione

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Novel pentasubstituted 1,8-naphthyridin-2-thione derivatives **9** have been isolated in high yield via a facile one-pot reaction of 3-arylidene-pent-2,4-dione derivatives **3** and cyanothio-acetamide **4**. Plausible mechanistic pathways are presented to account for the unexpected product. The new compounds were established via IR, ${}^{1}H_{-}{}^{13}C$ NMR, and MS analysis.

Keywords Cyanothioacetamide; mass fragment; mechanism; pentasubstituted 1,8-naphthyridin-2-thiones; spectra

2-Cyanothioacetamide is a highly reactive and readily obtainable starting material that has been extensively utilized in heterocyclic synthesis. ^{1–5} The reaction of 2-cyanothioacetamide with 1,3-diketones mainly afforded pyridine-2-thiones. ^{4,5} Annelated pyridine-2-thiones have shown remarkable pharmaceutical, ^{6–8} and biological activities. ^{9,10} Shuttleworth reported a parallel solution synthesis of pyridine-2-thione core via reaction of cyanothioacetamide with alkylated 1,3-diketones in the presence of excess of 4-methylmorpholine, as is shown in Scheme 1. Synthesis of similar pyridine-2-thiones core using the same starting materials has been recently reported. ¹² In continuation with our previous interest in the synthesis of azaheterocycles, ^{13–17} we report herein different and unexpected results for the reaction of cyanothioacetamide with the well-known 3-arylidenacetylacetone derivatives under the same conditions.

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 $R' = CO_2Me$, O-Ph, Me, $(CH_2)_4Me$, F,Cl-C₆H₃-CH₂-

SCHEME 1

RESULTS AND DISCUSSION

Our initial strategy aimed to synthesize 5-benzylidene-4,6-dimethyl-3-cyano-pyridine-2-thione **5** core which assumed to be a good synthon in a further heterocyclic synthesis, as is shown in Scheme 2. The benzylidene acetylacetone **3** easily reacted with cyanothioacetamide **4** in ethanol containing piperidine to yield yellow crystals after about 30 min of reflux. The IR spectrum of that product revealed intense absorption bands at v 1572, 1698, 2218, 3292, and 3377 cm $^{-1}$. The mass spectra (70 eV) showed molecular ion at 335 (M+1, 1), 334 (M $^+$, 3). The 1 H NMR spectrum (DMSO- d_6) showed signals at δ 13.7 (1H), 7.1–7.3 (5H), 4.9–4.4 (2H), 2.5 (3H), 2.1 (3H). Since our expectation was that product would be 5-benzylidene-3-cyano-4,6-dimethylpyridine-2-thione **5** which has m/z = 252 corresponds to molecular formula $C_{15}H_{12}N_2S$. Thus, this assumption is ruled out based on obtained spectral and elemental data, (Scheme 2).

SCHEME 2

The presence of intense absorption bands in the ir spectrum due to carbonyl and amine functions, as well as the presence of two singlet signals in the ¹H-NMR each integrating for three protons of two methyl groups, prompted us to suggest a plausible mechanistic pathway as is shown in Scheme 3. The addition of the activated methylene of cyanothioacetamide 4 into the activated double bond of arylidene 3 afforded

$$\begin{array}{c} \text{CH}_{3} & \text{Ar} \\ \text{O} & \text{H}_{2}\text{N} \\ \text{S} \\ \text{3a-d} \\ \text{a: Ar} = \text{C}_{6}\text{H}_{5} \\ \text{b: Ar} = p\text{-NO}_{2}\text{-C}_{6}\text{H}_{4} \\ \text{c: Ar} = p\text{-CH}_{3}\text{O-C}_{6}\text{H}_{4} \\ \text{d: Ar} = p\text{-CH}_{3}\text{O-C}_{6}\text{H}_{4} \\ \text{d: Ar} = p\text{-CH}_{3}\text{O-C}_{6}\text{H}_{4} \\ \text{d: Ar} = p\text{-CH}_{3}\text{O-C}_{6}\text{H}_{4} \\ \text{O} & \text{H}_{3}\text{C} & \text{N} \\ \text{S} \\ \text{N} & \text{S} \\ \text{H}_{3}\text{C} & \text{N} \\ \text{S} \\ \text{H}_{2}\text{N} & \text{S} \\ \text{O} & \text{H}_{2}\text{N} \\ \text{O} & \text{H}_{3}\text{C} \\ \text{N} & \text{S} \\ \text{Ar} & \text{NH}_{2} \\ \text{O} & \text{O} & \text{O} \\ \text{H}_{3}\text{C} \\ \text{N} & \text{S} \\ \text{H}_{3}\text{C} \\ \text{N} & \text{S} \\ \text{Ar} & \text{NH}_{2} \\ \text{O} & \text{O} & \text{O} \\ \text{H}_{3}\text{C} \\ \text{N} & \text{S} \\ \text{Ar} & \text{NH}_{2} \\ \text{O} & \text{O} & \text{O} \\ \text{H}_{3}\text{C} \\ \text{N} & \text{S} \\ \text{Ar} & \text{NH}_{2} \\ \text{O} & \text{O} & \text{O} \\ \text{H}_{3}\text{C} \\ \text{N} & \text{S} \\ \text{Ar} & \text{NH}_{2} \\ \text{O} & \text{O} & \text{O} \\ \text{H}_{3}\text{C} \\ \text{N} & \text{S} \\ \text{Ar} & \text{NH}_{2} \\ \text{O} & \text{O} & \text{O} \\ \text{H}_{3}\text{C} \\ \text{N} & \text{N} \\ \text{H}_{3}\text{C} \\ \text{N} \\ \text{N} \\ \text{H}_{3}\text{C} \\ \text{N} \\ \text{N} \\ \text{N} \\ \text{H}_{3}\text{C} \\ \text{N} \\ \text{N} \\ \text{H}_{3}\text{C} \\ \text{N} \\ \text$$

SCHEME 3

To confirm this assumption, possible fragmentation pathways have been proposed for the arylidene $\bf 3a$ and the product $\bf 9a$, as is shown in Charts 1 and 2, respectively. On the other hand, several similar structures, each with m/z=335/336, can be constructed as possible products, however, they are ruled out based on spectral data (Schemes 4 and 5). The cyclic dithione intermediate $\bf 11$ may result from a self-condensation of cyanothioacetamide $\bf 4$, followed by condensation with the benzylidene $\bf 3$ to yield the intermediate $\bf 12$. This intermediate $\bf 12$ may cyclizes to 1,4-naphthyridine-2,8-dithione $\bf 13$ via loss of water. Based on the presence

SCHEME 4

CHART 1

of the carbonyl absorption band at v 1698 cm⁻¹, and a singlet signal due to NH₂ at δ 4.4 ppm, this assumption is ruled out (Scheme 4).

Similarly, the cyanothioacetamide **4** could react with the pyridine **5** via losing H_2S to yield the proposed 1,8-naphthyridine **14**, which may hydrolyze into its carboamide derivative **15** which has the molecular formula $C_{18}H_{16}ON_4S$ (336). Also, this pathway could safely be excluded since there is an intense absorption band at v 2218 cm⁻¹ assigned for nitrile function, and only one singlet signal due to two protons of amine function (Scheme 5).

EXPERIMENTAL

All melting points were uncorrected. The ${}^{1}\text{H}$ -, ${}^{13}\text{C NMR}$ spectra (DMSO- d_{6} , δ in ppm) were run on a Varian EM-390 spectrometer using

CH₃ Ph NH₂ CN
$$\frac{1}{1}$$
 NH₃ N

CHART 2

TMS as internal standard at Dusiburg-Essen University, Germany. Their spectra (KBr, ν in cm⁻¹) were recorded on a Pye-Unicam SP-1100 Spectrophotometer. Mass spectra were recorded on a Varian MAT 311 A spectrometer at 70 eV, and the elemental analysis were determined at the Microanalytical Center, Cairo University, Egypt.

NC
$$S = NH_2$$
 $S = NH_2$ $S = NH$

SCHEME 5

3-Arylideneacetylacetone derivatives **3a–d** were synthesized according to the method of Horning et al. ¹⁸

Preparation of 6-Acetyl-4-amino-5-aryl-3-cyano-7-methyl-1,8-naphthyridin-2-thiones (9a-d)

To 50 ml of ethanol containg $3.8~\mathrm{g}$ (2 mmol) of 3a and $2.0~\mathrm{g}$ (2 mmol) of cyanothioacetamide 4 an $0.2~\mathrm{ml}$ of piperidine was added, and the mixture was refluxed. The yellow needles formed after $15~\mathrm{minutes}$ was filtered from the solution after one hour and recrystallized from a mixture of ethanol-dimethyl-formamide (3:1) into yellow needles. In analogy, compounds 3b-d reacted with cyanothioacetamide $4~\mathrm{to}$ give the products 9b-d, respectively.

6-Acetyl-4-amino-3-cyano-7-methyl-5-phenyl-1,8-naphthyridin-2-thione (9a)

The yield in this mixture is 4.0 g (59%); mp: 310–312 °C; IR: 1572 (C=S), 1698 (C=O), 2218 (CN), 3292 and 3377 (NH and NH₂); ¹H NMR: 13.7 (s, 1H, NH), 7.07–7.30 (m, 5H, Ar-H), 4.9–4.4 (br, 2H, NH₂), 2.5 (s, 3H, CH₃), 2.1 (s, 3H, CH₃); ¹³C NMR: 209.3 (C=S), 175.0 (C=O), 155.2 (*C*-8a), 152.5 (*C*-7), 143.7 (*C*-4), 128.9–123.9 (phenyl-*C*, *C*-3, *C*-4a, *C*-5 and *C*-6), 116.5 (CN), 27.7 (CH₃–CO), 18.6 (CH₃). MS, m/z (%): 335 (M + 1, 1), 334 (M⁺, 2), 309 (2), 291 (100), 276 (8), 233 (3), 189 (3), 165 (4), 151 (3). Anal. Calcd. for C₁₈H₁₄N₄OS (334.40): C, 64.65; H, 4.22; N, 16.75; S, 9.59. Found: C, 64.46; H, 4.10; N, 16.56; S, 9.44.

6-Acetyl-4-amino-3-cyano-7-methyl-5-(4-nitrophenyl)-1,8-naphthyridin-2-thione (9b)

The yield for this mixture is 2.5 g (66%); mp: 190–192°C (MeOH); IR: 1575 (C=S), 1695 (C=O), 2215 (CN), 3290 and 3372 (NH and NH₂); $^1\mathrm{H}$ NMR: 13.2 (s, 1H, NH), 7.1–7.7 (m, 4H, Ar-H), 4.5 (s, 2H, NH₂), 2.4 (s, 3H, CH₃), 1.9 (s, 3H, CH₃); $^{13}\mathrm{C}$ NMR: 208.5 (C=S), 176.1 (C=O), 152.4 (*C-8a*), 150.5 (*C-7*), 144.2 (*C-4*), 138.2 (phenyl-4-NO₂), 131.2–123.7 (phenyl-C, *C-3*, *C-4a*, *C-5* and *C-6*), 117.3 (CN), 27.3 (CH₃–CO), 18.4 (CH₃). MS, m/z (%): 381 (M + 2, 2), 380 (M + 1, 4), 379 (M⁺, 55). Anal. Calcd. for C₁₈H₁₃N₅O₃S (379.40): C, 56.98; H, 3.45; N, 18.46; S, 8.45. Found: C, 56.81; H, 3.26; N, 18.28; S, 8.27.

6-Acetyl-4-amino-3-cyano-7-methyl-5-(4-chloro)phenyl-1,8-naphthyridin-2-thione (9c)

The yield for this mixture is 2.0 g (54%); mp: 240–242°C (MeOH); IR: 1570 (C=S), 1694 (C=O), 2218 (CN), 3291 and 3375 (NH and NH₂); ¹H NMR: 12.8 (s, 1H, NH), 7.1–7.5 (m, 4H, Ar-H), 4.7 (s, 2H, NH₂), 2.3 (s, 3H, CH₃), 2.2 (s, 3H, CH₃); ¹³C NMR: 207.8 (C=S), 175.6 (C=O), 154.5 (C-8a), 151.3 (C-7), 141.5 (C-4), 135.2 (Phenyl-4-Cl), 128.9–122.9 (phenyl-C, C-3, C-4a, C-5 and C-6), 118.1 (CN), 27.1 (CH₃–CO), 16.3 (CH₃). MS, m/z (%): 370 (M + 2, 3), 369 (M + 1, 2), 368 (M⁺, 65). Anal. Calcd. for C₁₈H₁₃N₄OSCl (368.84): C, 58.62; H, 3.55; N, 15.19; S, 8.69; Cl, 9.61. Found: C, 58.47; H, 3.38; N, 15.01; S, 8.52; Cl, 9.45.

6-Acetyl-4-amino-3-cyano-7-methyl-5-(4-methoxy)-phenyl-1,8-naphthyridin-2-thione (9d)

This mixture was precipitated after 20 min of reflux as yellow crystals. The yield is 2.8 g (76%); mp: 295–296°C (DMF/EtOH); IR: 1570 (C=S), 1694 (C=O), 2218 (CN), 3291 and 3375 (NH and NH₂); $^1\mathrm{H}$ NMR: 13.6 (s, 1H, NH), 7.1–7.7 (m, 4H, Ar-H), 4.7 (s, 2H, NH₂), 3.6 (s, 3H, OCH₃), 2.4 (s, 3H, CH₃), 2.2 (s, 3H, CH₃); $^{13}\mathrm{C}$ NMR: 208.5 (C=S), 175.3 (C=O), 154.8 (*C*-8a), 153.9 (*C*-7), 142.1 (*C*-4), 132.9–123.4 (phenyl-C, *C*-3, *C*-4a, *C*-5 and *C*-6), 117.3 (CN), 55.4 (OCH₃) 26.8 (CH₃–CO), 17.4 (CH₃). MS, m/z (%): 366 (+2, 2), 365 (M + 1,1), 364 (M⁺, 75). Anal Calcd. for C₁₉H₁₆N₄O₂S (364.42): C, 62.62; H, 4.43; N, 15.37; S, 8.80. Found: C, 62.47; H, 4.28; N, 15.21; S, 8.62.

REFERENCES

- [1] G. E. H. Elgemeie and A. H. Elghandour, Bull. Chem. Soc. Jpn., 63, 1230 (1990).
- [2] G. E. H. Elgemeie, A. M. Elzanate, and A. K. Mansour, J. Chem. Soc., Perkin Trans., 1, 1073 (1992).

- [3] G. E. H. Elgemeie, I. S. Alnaimi, and H. F. Alarab, Heterocycles, 34, 1721 (1992).
- [4] G. E. H. Elgemeie, H. A. Ali, and M. M. Eid, J. Chem. Res. (S), 256 (1993).
- [5] G. E. H. Elgemeie, H. A. Elfahham, and H. A. Nabey, Bull. Chem. Soc. Jpn., 61, 4431 (1988).
- [6] K. Hirai, Y. Iwano, T. Nishi, A. Youshida, K. Oda, and H. Koyama, U.S. Pat 5. 541, 317; C. A. 125, 167960 (1996).
- [7] Y. Momose and H. Odaka, PCT Int. Appl. WO 97 36, 882; C. A., 117, 331478 (1997).
- [8] Y. Youshida, H. Kogen, I. Hayakawa, et al., JP 09, 292, 775; C. A., 127, 234181 (1997).
- [9] J. K. Hinks, A. K. Takel, and E. Hunt, PCT Int. Appl. WO 9725, 309; C. A., 127, 161997 (1997).
- [10] L. Muthusubramanian, R. R. Mitra, S. Rajkumar, and V. S. S. Rao, J. Chem. Technol. Biotechnol., 72, 164 (1998).
- [11] S. J. Shuttleworth, M. Quimpere, N. Lee, and J. DeLuca, Molecular Diversity, 4, 183 (1998).
- [12] M. A. A. Elneary, Phosphorus, Sulfur and Silicon, 178, 2201 (2003).
- [13] S. M. Sayed, M. A. Khalil, M. A. Ahmed, and M. A. Raslan, Synth. Communi., 32(3), 481 (2002).
- [14] F. M. Abd El Latif, E. El-Rady, M. A. Khalil, and M. A. El-Maghraby, J. Heterocyclic Chem., 39, 299 (2002).
- [15] M. A. Raslan, S. M. Sayed, M. A. Khalil, and A. M. Farag, *Heteroatom Chem.*, 11(2), 94 (2000).
- [16] S. M. Sayed, M. A. Selim, M. A. Raslan, and M. A. Khalil, *Heteroatom Chem.*, 11(5), 362 (2000).
- [17] M. A. Khalil, J. Chin. Chem. Soc., 49(6), 1069 (2002).
- [18] E. C. Horning, J. Koo, M. S. Fish, and G. N. Walker, Organic Synthesis (Wiley, New York, 1983), Vol. 4, 408.